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e Jet	1. 50X1	The following Chemistry In Budapest	ng research was in natitute at the Eoe	progress voes Lo	in the Phyrand Univer	ysical esity of		
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9 3		face. on meth	uring the polarizat This involved search ods of stopping the y surface treatment	ion capa ing for corrosi	city of the additional on. perhaps	e metal sur- l information s with a pre-	ı	
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(c)	Measurement of polarization capacity. The aim was to find
(-)	a method to accurately determine the polarization capacity
	and in this connection an absolute "measurement" of metal
	surfaces, which involved examination of adsorption as well
	as investigation of corrosion. This research was done by
50X1	two research assistants and a part time laboratory assistant
50X1	The research assistants were Jozsef Devay, and
	Mrs Bela Beress, nee Maria Triznyai
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Anodic polishing of copper and iron. The aim was to develop an economical industrial process, if possible by use of domestic materials (acids), and the study of the reaction mechanism of electrode processes. This research was done by one adjunktus (associate professor), and two research assistants, Mrs Andras Szokoby, nee Dr Ilona Botka, Endre Barla, 50X1 50X1 Endre Barla, 50X1 Mrs Karoby Urhegyi, nee Marta Vanyek

(e) Examination of the mechanism of the migration of postive hydrogen ion based on the Grotthus theory; measurement of transport number of H ions in an alcohol-water mixture of different concentrations by the moving boundary method (also in absolute alcohol medium). Further measurements were planned using glycerin-water and other water-compound (containing HO-group) mixtures. Measurements of the conductivity and viscosity paralleled the transport number measurements. research assistants and one laboratory assistant carried out this research; Lajos Majtenyi and Miss Antonia The measurements were analogous from the viewpoint of measurement technique with those carried out in the laboratory directed by A R Gordon in Toronto, Canada.

Diffusion tests, which since 1937 had been done by the diaphragm method, were being continued. These tests were done by Andor Hunyar in 1938, by Eva Pogany in 1943, by Tereze Schandl in

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- 1948, and in 1950 in diffusion of sugar. The dissertation of Klara Arkosi, 1946, 50X1 and Alajos Vali, 1945, dealt with the distribution of pores in sintered glass filters and their transmissability. Hildegard Hartmann, in March 1950, measured the change of the diffusion constant of different electrolytes as a function of the electrolyte concentration (analogous to the method used to attain results published by R H Stokes in the "Journal of the American Chemical Society", May 1950).
- Tests in selective catalytic hydrogenation in the presence of Pd as satalyzer on the surface of BaSO4. Experiments were made to see whether a catalyzer of identical properties could be reproduced. This included the study of the mechanism of reaction, the study of the dependence of the catalytic effect on the solvent, and the pH, and the study of the effect of the dispersion of the open of the catalytic effect of the dispersion of the pH, and the study of the effect of the dispersion of the pH. sion grade of the catalyzer. Research was done by Kroly Zimmer, research assistant and Laszlo Kiss, research assistant, and one laboratory assistant.

(h) Examination of gas absorption in solutions in initial stages. Research was done by Dr Elvina Kugler, assistant.

Determination of surface by means of gas adsorption measurements. This experiment was done by the method of Brounauer-Emmett. The original aim of the examination was the absolute surface determination of platinum black, but the method was not sensitive

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enough for this. This method can be used for the surface measurement of powder catalyzers. Determination on this basis of dispersion makes it possible to compare the results with other surface measurements. Research was done by Mrs Ferenc Nagy, nee Agnes Sipos

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(j) Production of silica gel of different dispersion grades which was reproducible for catalytic purposes, and which could be used industrially. Research was done by Dr Mrs Mihaly Minesev, dozent, and one laboratory assistant.

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